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Synthesis of Silicon-Containing α -Amino Acids and Hydantoins

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Abstract: Three new silicon-containing α -amino acids 6a-c and their respective hydantoin derivatives 8a-c have been prepared in racemic form. Alkylation of an α -amino cyanoacetate derivative with iodomethyl-substituted silanes, followed by hydrolysis gave the amino acids, and the subsequent reactions with KOCN and with HCl gave the hydantoins. The hydantoins are planned to be subjected to a biotechnological deracemization process. © 1997 Elsevier Science Ltd.

INTRODUCTION

There is a worldwide increase of interest in the synthesis and the use of optically pure natural and unnatural α -amino acids.² Such compounds are principally available by three processes: by enantioselective synthesis, by resolution, or by deracemization. The last of the three processes is of particular interest since it would allow to use the chemically most simple synthetic pathways towards racemic compounds and would still permit to obtain the desired enantiomerically pure substrates in theoretically quantitative yields. Syldatk *et al.* have developed such a deracemization process based on a combination of biotransformations:³ they transferred racemic hydantoins 1 efficiently into either of the two antipodes of the respective α -amino acids 2 by use of three synergetically acting enzymes: a hydantoin racemase — providing rapid interconversion of the enantiomeric hydantoins L-1 and D-1 into each other — and a combination of L- or D-selective hydantoinases

Scheme 1

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and carbamoylases — responsible for the enantiomer-selective degradation of either the L- or the D-hydantoins 1 to the respective enantiomerically pure α-amino acids (Scheme 1). A number of optically pure unusual and unnatural α-amino acids have been prepared by this "hydantoinase" process so far. The method has also been applied to the synthesis of D-3-(trimethylsilyl)alanine (D-TMS-Ala),⁴ a compound that was successfully used as a bioisoster for phenylalanine in renin inhibitors.⁵ Since silylated amino acids others than TMS-Ala could be of substantial interest as substitutes for natural lipophilic amino acids, too, we planned to use the "hydantoinase" process for the preparation of the optically active amino acids D-6a-c and L-6a-c. In the following we describe the synthesis of the racemic precursor compounds 6a-c and their respective hydantoin derivatives 8a-c.

RESULTS AND DISCUSSION

For the synthesis of the compounds 8a-c, we followed a reaction sequence that was earlier used for the preparation of TMS-substituted hydantoin 8d (Scheme 2). Treatment of cyanoacetate derivative 3 with the iodomethyl-substituted silanes 4a-c^{11,12} in the presence of base provided the respective alkylated derivatives 5a-c in moderate yields. These compounds were hydrolyzed with aqueous NaOH and gave rise to the amino acids 6a-c after decarboxylation of the intermediary malonic acid derivatives. It is crucial for the hydrolysis step to use at least three equivalents of base and to heat the reaction mixture for a prolonged period of time (usually 12-48 h). The use of less NaOH or shorter reaction times gave incomplete transformations. As a result, considerable amounts of N-acetyl amino acids were isolated. Conversion of the amino acids 6a-c to the corresponding hydantoins 8a-c was attained by their treatment with KOCN in pyridine/H₂O and the subsequent cyclization of the crude N-carbamoylamino acids 7a-c with HCl. The overall yields of the hydantoins 8a-c summed up to approximately 20-23% (10% for the phenyl-substituted compound 8c) when the reaction

sequences were performed stepwise with isolation and purification of the intermediary amino acids 6a-c. When purification of these intermediates was omitted, the yields increased to 40-48% (20% for 8c). The significantly lower amounts that were obtained in the preparation of the (dimethyl)phenylsilyl-substituted compound 8c was attributed to the fairly labile Ph-Si bond. Particularly in the hydrolysis step, under the prolonged action of hydroxide base, an ample amount of insoluble material was formed, and the yield of the desired product decreased drastically.

The synthetic pathway towards silicon-containing α-amino acids and hydantoins presented in this paper is straightforward and leads to the desired products — except for 6c/8c — in reasonable overall yields. Problematic in the reaction sequence are two steps: the initial alkylation reaction (formation of 5a-c from 3 and 4a-c) and in the preparation of the phenyl-substituted silicon compound 6c, as already mentioned above, the hydrolysis step. The alkylation reaction seems to be difficult mostly for steric reasons: with smaller and more reactive electrophiles, like, e.g., MeI but also with Me₃SiCH₂I, the yields were considerably higher (up to 90% as compared to approximately 45%). We have also tested alternative glycine enolate equivalents (enolates of N-benzylidene glycine or of phthalimido-protected glycine) that thus gave inferior results.

EXPERIMENTAL

General Remarks. Unless otherwise stated: all org. solvents were distilled prior to use. All moisture-sensitive reactions were carried out under a blanket of Ar. Extracts were dried (MgSO₄) and evaporated in vacuo. Column chromatography (SiO₂): silica gel Merck 60 (40–63 µm). M.p.: Mettler FP-5/FP-52; d = decomposition. IR: Perkin-Elmer 781; as KBr pellets; data in ∇ (cm⁻¹) with s (strong), m (middle), and w (weak) with respect to the most intense absorption. ¹H NMR: at 300 MHz in CDCl₃; Bruker AC-300 or Bruker ARX-300; δ in ppm relative to CHCl₃ (δ = 7.26) or DMSO (δ = 2.49); J in Hz; multiplicities denoted as s = singlet, d = doublet, t = triplet, q = quadruplet, quint. = quintuplet, and m = multiplet. ¹³C NMR: at 75.6 MHz in CDCl₃; Bruker ARX-300; δ in ppm relative to CDCl₃ (δ = 77.0) or DMSO (δ = 39.7); multiplicities from DEPT90 and DEPT135 experiments. CI-MS (chemical ionization mass spectrometry) with NH₃ as the reactant gas: Finnigan SSQ 700 or Finnigan MAT 90; ESI-MS (electrospray ionization mass spectrometry): Finnigan TSQ 700; signals at m/z > 50 and >10% rel. abundance, for ESI-MS: base peak and quasi-molecular ions only; data in m/z (relative abundance of major isotope in percent, where appropriate). Elemental analyses: Elementar Vario EL.

Ethyl 2-Acetamido-2-cyano-3-(1-methyl-1-silacyclohex-1-yl)propionate (5a): A soln. of ethyl N-acetamidocyanoacetate (3, 1.17 g, 6.88 mmol) in DMSO (10 ml) was added at 23°C to a soln. of NaOEt (0.47 g, 6.86 mmol) in DMSO (10 ml). After 4 h, 1-(iodomethyl)-1-methylsilacyclohexane ¹¹ (4a, 1.47 g, 5.80 mmol) was added dropwise and the resultant mixture was stirred at 23°C for 48 h. It was quenched with H₂O, extracted with Et₂O, and chromatographed (AcOEt/hexane 1:1) to give 5a (0.85 g, 2.87 mmol, 49%) as colorless crystals. M.p.: 97.8–98.6°C (AcOEt/hexane). IR: 3270s, 3190m, 3020m, 2990m, 2910s, 2850m, 2790w, 1760s, 1710w, 1695m, 1660s, 1535s, 1480w, 1460w, 1445m, 1405m, 1390m, 1375m, 1360m, 1340w,

1300s, 1265s, 1205s, 1180s, 1160w, 1125w, 1115w, 1095w, 1020s, 995m, 970w, 910s, 870w, 845s, 820s, 795s, 780m, 760w, 705w, 685m, 645w. 1 H NMR: 6.21 (s, HN); 4.36, 4.26 (2dq, J = 10.7, 7.1, MeCH₂); 2.05 (s, MeCO); 1.77, 1.49 (AB, J = 14.6, SiCH₂); 1.75–1.40 (m, CH₂CH₂CH₂); 1.37 (t, J = 7.1, $MeCH_2$); 0.67 (q-like m, 'J' = 7.1, 2 SiCH₂); 0.18 (s, MeSi). 13 C NMR: 169.3 (s, MeCO); 167.3 (s, CO₂Et); 117.3 (s, C=N); 63.7 (t, OCH₂); 54.5 (s, CC=N); 29.6 (t, Si(CH₂)₂ CH_2); 24.0 (t, 2 SiCH₂ CH_2); 23.8 (t, SiCH₂); 22.6 (t, t), t0, t1, t2, t3, t4, t5, t6, t7, t7, t8, t8, t9, t9,

Ethyl 2-Acetamido-2-cyano-3-(1-methyl-1-silacyclopent-1-yl)propionate (5b): Analogously to the preparation of 5a, 3 (3.13 g, 18.4 mmol) was deprotonated with NaOEt (1.25 g, 18.4 mmol) and alkylated with 1-(iodomethyl)-1-methylsilacyclopentane¹¹ (4b, 3.72 g, 15.5 mmol). The solid obtained after work-up was recrystallized (AcOEt) and gave 5b (2.00 g, 7.08 mmol, 46%) as colorless crystals. M.p.: 129.8–131.5°C (AcOEt). IR: 3260s, 3030m, 3000m, 2930s, 2850m, 2800w, 1760s, 1710m, 1660s, 1650s, 1550m, 1540s, 1505m, 1480m, 1450m, 1405m, 1390m, 1375s, 1360m, 1300s, 1270s, 1200s, 1155m, 1115m, 1090m, 1075s, 1020s, 960w, 925w, 865s, 835s, 805s, 785s, 775s, 710m, 670m, 655m. 1 H NMR: 6.21 (s, HN); 4.36, 4.26 (2dq, $J = 10.7, 7.1, MeCH_2$); 2.05 (s, MeCO); 1.78, 1.58 ($AB, J = 14.6, SiCH_2$); 1.61–1.54 (m, 2 SiCH₂CH₂); 1.37 (t, $J = 7.1, MeCH_2$); 0.70–0.56 (m, 2 SiCH₂); 0.26 (s, MeSi). 13 C NMR: 169.3 (s, MeCO); 167.1 (s, CO_2EI); 117.3 (s, C=N); 63.8 (t, OCH₂); 54.6 (s, CC=N); 26.9, 26.8 (2t, 2 SiCH₂CH₂); 24.5 (t, SiCH₂); 22.6 (q, MeCO); 13.8 (q, $MeCH_2$); 12.4, 12.3 (2t, 2 SiCH₂); -2.5 (q, MeSi). CI-MS: 300 (100, $[M + NH_4]^+$), 283 (20, $[M + H]^+$), 256 (7), 242 (13). Anal. Calcd. for $C_{13}H_{22}N_2O_3Si$ (282.42): C 55.29, H 7.85, N 9.92. Found: C 54.76, H 7.77, N 10.03.

3-(1-Methyl-1-silacyclohex-1-yl)alanine (6a): A soln. of **5a** (0.77 g, 2.60 mmol) and NaOH (2.0 g, 50.0 mmol) in H₂O (20 ml) was refluxed for 12 h. It was cooled to 0°C and acidified with conc. aq. HCl soln. to pH = 6. The thus formed precipitate was filtered off and chromatographed (CH₂Cl₂/EtOH/AcOH 14:4:1) to give **6a** (0.45 g, 2.23 mmol, 86%) as colorless crystals. M.p.: 239.8–241.3(d)°C (H₂O). IR: 3400*m*, 3000*s*, 2910*s*, 2850*s*, 2610*m*, 2520*m*, 1745*s*, 1590*m*, 1500*m*, 1445*m*, 1400*w*, 1345*w*, 1250*s*, 1205*m*, 1180*s*, 1160*s*, 1075*w*, 1025*w*, 990*m*, 910*m*, 860*m*, 810*s*, 800*s*, 755*m*, 720*m*, 705*w*, 675*w*. ¹H NMR (DMSO/DCl): 3.69 (*dd*, J = 12.1, 4.1, CH); 1.68–1.40 (*m*, 2 SiCH₂CH₂); 1.41–1.22 (*m*, Si(CH₂)₂CH₂); 1.17, 1.01 (*AB* of *ABX*, $J_{AB} = 14.3$, $J_{AX} = 4.2$, $J_{BX} = 12.2$, SiCH₂); 0.65–0.45 (*m*, 2 SiCH₂); 0.01 (*s*, MeSi). ¹³C NMR (DMSO/DCl): 171.7 (*s*, CO₂H); 50.2 (*d*, CH); 29.5 (*t*, Si(CH₂)₂CH₂); 24.0, 23.9 (2*t*, 2 SiCH₂CH₂); 16.8, 13.1, 12.7 (3*t*, 3 SiCH₂); -4.4 (*q*, MeSi). CI-MS: 202 ([*M* + H]+). Anal. Calcd. for C₉H₁₉NO₂Si (201.34): C 53.69, H 9.51, N 6.96. Found: C 53.22, H 9.22, N 6.83.

3-(1-Methyl-1-silacyclopent-1-yl)alanine (6b): Analogously to the preparation of **6a**, **5b** (1.9 g, 6.73 mmol) was hydrolyzed with NaOH (2.0 g, 50.0 mmol) and gave, after recrystallization of the crude product (H₂O), **6b** (0.92 g, 4.89 mmol, 73%) as colorless crystals. M.p.: 252(d)°C (H₂O). IR: 3420m, 3040s, 2920s, 2850s, 1675s, 1600s, 1550s, 1430s, 1395s, 1345m, 1320m, 1245s, 1170s, 1130s, 1075s, 1025s, 885w, 860m, 830s, 775s, 740m, 705m, 650m, 635w. H NMR (DMSO/DCl): 3.66 (dd, J = 12.4, 4.2, CH); 1.24 (s, 2 SiCH₂CH₂); 1.11, 0.91 (dB of dBX, dB = 14.1, dA = 4.2, dB = 13.3, SiCH₂); 0.33–0.20 (dB = 2 SiCH₂CH₂); -0.08 (dB = 13.5 (dB = 14.1, dB = 15.3, dB = 15.3,

3-[(Dimethyl)phenylsilyl]alanine (6c): Analogously to the preparation of **6a**, **5c** (0.62 g, 1.93 mmol) was hydrolyzed with NaOH (2.0 g, 50.0 mmol) and gave, after recrystallization of the crude product (H₂O), **6c** (0.30 g, 1.35 mmol), 70%, contaminated with inorganic impurities) as a colorless amorphous solid. M.p.: 213.7-215.3(d)°C (H₂O). IR: 3420m, 3060s, 1610s, 1515m, 1505m, 1485m, 1470w, 1455w, 1425s, 1415m, 1390s, 1345m, 1315m, 1245s, 1180m, 1140w, 1115m, 1020w, 995w, 885w, 835s, 810m, 780s, 740m, 725s, 700s, 665w, 635w, 620w. ¹H NMR (DMSO/DCl): 7.48-7.45 (m, 2 arom. H); 7.33-7.30 (m, 3 arom. H); 3.73 (dd, J = 12.0, 4.1, CH); 1.40, 1.26 (AB of ABX, $J_{AB} = 14.4$, $J_{AX} = 4.2$, $J_{BX} = 12.3$, SiCH₂); 0.27, 0.24 (2s, Me₂Si). ¹³C NMR (DMSO/DCl): 172.3 (s, CO₂H); 139.0 (s, arom. C); 134.9 (d, 2 arom. C); 130.8 (d, arom. C); 129.5 (d, 2 arom. C); 51.7 (d, CH); 19.6 (t, SiCH₂); -1.1, -1.5 (2q, Me₂Si). ESI-MS (MeOH): 246 ([M + Na]+), 224 ([M + H]+).

N-Carbamoyl-3-(1-methyl-1-silacyclohex-1-yl)alanine (7a) and 5-[(1-Methyl-1-silacyclohex-1-yl)methyl]imidazolidine-2,4-dione (8a): From 6a: A suspension of 6a (0.20 g, 1.0 mmol) and KOCN (0.10 g, 1.2 mmol) in H₂O/pyridine (1:1, 4 ml) was heated to 60°C for 2 h (the reactants were dissolved completely). It was cooled to 23°C, pyridine was extracted with AcOEt, and the aq. layer acidified with aq. HCl soln. (6N, 1 ml) and AcOH (1 ml) (formation of a precipitate). The mixture was heated to reflux for 1 h, cooled to 4°C and stored at this temperature for 15 h to complete crystallization of the desired product. The precipitate was filtered off and dried to give 8a (0.12 g, 0.55 mmol, 55%). The N-carbamoyl intermediate 7a was ob-

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tained, when the intermediary formed precipitate (after treatment of 6a with KOCN and acidification) was filtered off, washed with H₂O, and dried. From 3: Analogously to the preparation of 5a, 3 (14.36 g, 84.4 mmol) was deprotonated with NaOEt (6.7 g, 92.8 mmol) and alkylated with 4a (18.64 g, 73.3 mmol). Analogously to the preparation of 6a, the crude 5a was directly hydrolyzed with NaOH (20 g, 0.50 mol, in 200 ml of H₂O, reflux for 48 h), and the thus obtained crude amino acid was subsequently treated as above with KOCN (7.3 g, 90 mmol) in H₂O/pyridine (1:1, 200 ml) and then with aq. HCl soln. (6N, 40 ml) and AcOH (40 ml) at reflux. Recrystallization (H₂O) of the crude product afforded 8a (8.0 g, 35.3 mmol, 48%) as colorless crystals.

Data of **7a**: M.p.: 123(d)°C (H₂O). IR: no reasonable spectrum was obtained. ¹H NMR (DMSO): 5.92 (d, J = 6.7, HN); 5.39 (s, H₂N); 3.72 (q, J = 6.8, CH); 1.68–1.42 (m, 2 SiCH₂CH₂); 1.39–1.20 (m, Si(CH₂)₂CH₂); 1.01, 0.89 (AB of ABX, J_{AB} = 14.3, J_{AX} = 6.9, J_{BX} = 6.8, SiCH₂); 0.67–0.42 (m, 2 SiCH₂); 0.00 (s, MeSi). ¹³C NMR (DMSO): 176.5, 158.3 (2s, 2 C=O); 51.9 (d, CH); 30.0 (t, Si(CH₂)₂CH₂); 24.4 (t, 2 SiCH₂CH₂); 20.3, 13.8, 13.7 (3t, 3 SiCH₂); –3.8 (q, MeSi). CI-MS (compound decomposes): 30°C: 263 (100); 60°C: 279 (25), 263 (100); 150°C: 528 (19), 391 (20), 365 (35), 295 (8), 279 (100), 263 (13), 223 (4); 250°C: 536 (7), 462 (4), 367 (10), 314 (29), 279 (8), 255 (13), 227 (48), 202 (33), 184 (50), 158 (90), 145 (40), 132 (46), 130 (100), 114 (73). ESI-MS (MeOH): 479 (19), 463 (19), 369 (20), 338 (35), 327 (44), 311 (18), 283 (36), 245 (100, [M + H]⁺), 228 (34), 190 (75), 175 (68), 139 (48).

Data of **8a**: M.p.: 174.8–175.9°C (H₂O). IR: 3280m, 3180m, 3060m, 2910s, 2845m, 2790w, 2750w, 1775s, 1730s, 1695m, 1680w, 1645w, 1555w, 1535w, 1440m, 1415s, 1380m, 1335m, 1295m, 1255m, 1205m, 1185m, 1095m, 1040w, 1005w, 990m, 910m, 895m, 870m, 825s, 800s, 740m, 710w, 695w, 675w, 645m, 630w. ¹H NMR (DMSO): 10.53, 7.78 (2s, br., 2 HN); 4.08 (t, J = 6.3, CH); 1.62–1.23 (m, CH₂CH₂CH₂); 1.03, 0.94 (t and t and t and t are 14.8, t and 150 NMR (DMSO): 177.3, 157.3 (2s, 2 C=O); 55.1 (t and t are 150.10 (t and t are 160.10 (t and 160.10 (t

N-Carbamoyl-3-(1-methyl-1-silacyclopent-1-yl)alanine (7b) and 5-[(1-Methyl-1-silacyclopent-1-yl)methyl]imidazolidine-2,4-dione (8b): From 6b: Analogously to the preparation of 7a and 8a, 6b (0.75 g, 4.0 mmol) was reacted with KOCN (0.39 g, 4.8 mmol) in H₂O/pyridine (1:1, 16 ml) then with aq. HCl soln. (6N, 4 ml) and AcOH (4 ml) to form 8b (0.50 g, 2.35 mmol, 58%). N-Carbamoyl amino acid 7b was collected, as described above, after acid treatment before heating to reflux. From 3: Analogously to the preparation of 8a, 3 (11.41 g, 67.1 mmol) was deprotonated with NaOEt (4.56 g, 67.1 mmol) and alkylated with 4b (13.60 g, 56.5 mmol). It was hydrolyzed with NaOH (80 g, 2.0 mol, in 400 ml of H₂O, reflux 48 h), treated with KOCN (5.2 g, 64.0 mmol) in H₂O/pyridine (1:1, 200 ml), then with aq. HCl soln. (54 ml) and AcOH (54 ml). Recrystallization (H₂O) of the crude product afforded 8b (4.8 g, 22.5 mmol, 40%) as colorless crystals.

Data of 7b: M.p.: 146(d)°C (H₂O). IR: 3410s, 3380s, 3230s, 2935s, 2835m, 2480w, 1880w, 1690s, 1680s, 1660s, 1640s, 1560s, 1550s, 1540s, 1470m, 1450w, 1415m, 1405m, 1315m, 1295m, 1240s, 1200m, 1175m, 1150m, 1075s, 1050s, 1030s, 910w, 860m, 830m, 810s, 775m, 725w, 710w, 685w, 650w, 620w. ¹H

NMR (DMSO): 6.10 (d, J = 8.5, HN); 5.53 (s, H₂N); 4.08 (dt, J = 9.6, 6.0, CH); 1.49 (quint., J = 3.7, 2 SiCH₂CH₂); 1.06, 0.94 (AB of ABX, J_{AB} = 14.5, J_{AX} = 5.9, J_{BX} = 9.8, SiCH₂); 0.64–0.37 (m, 2 SiCH₂); 0.09 (s, MeSi). ¹³C NMR (DMSO): 176.0, 158.1 (2s, 2 C=O); 49.6 (d, CH); 26.8, 26.7 (2t, 2 SiCH₂CH₂); 19.4, 12.0, 11.9 (3t, 3 SiCH₂); -2.7 (q, MeSi). CI-MS: 248 (25, [M + NH₄]⁺), 231 (100, [M + H]⁺), 213 (10, [M + H - H₂O]⁺ \triangleq [8b + H]⁺), 188 (86, [M + H - CONH]⁺ \triangleq [6b + H]⁺).

Data of **8b**: M.p.: $163.5-164.2^{\circ}$ C (H₂O). IR: 3220s, 3060s, 2920s, 2850m, 2745w, 1770s, 1735s, 1695w, 1685w, 1555w, 1485w, 1450m, 1415s, 1380m, 1320m, 1305m, 1245m, 1205s, 1140s, 1070s, 1040s, 1005m, 945w, 890m, 860m, 835s, 805s, 785m, 760m, 740m, 715w, 700w, 650m, 635m. 1 H NMR (DMSO): 10.55, 7.81 (2s, br. 2 HN); 4.15-4.10 (m, CH); 1.49 (quint-like m, 'J' = 3.7, 2 SiCH₂CH₂); 1.09, 1.03 (AB of ABX, $J_{AB} = 14.8$, $J_{AX} = 6.4$, $J_{BX} = 6.0$, SiCH₂); 0.62-0.43 (m, 2 SiCH₂); 0.10 (s, MeSi). 13 C NMR (DMSO): 177.1, 157.3 (2s, 2 C=O); 55.3 (d, CH); 26.7 (t, 2 SiCH₂CH₂); 18.3, 12.4, 12.3 (3t, 3 SiCH₂); -2.3 (q, MeSi). CI-MS: 230 (100, [M + NH₄]⁺), <math>213 (17, $[M + H]^+$). Anal. Calcd. for C₉H₁₆N₂O₂Si (212.33): C 50.91, H 7.60, N 13.19. Found: C 50.72, H 7.33, N 12.98.

N-Carbamoyl-3-[(dimethyl)phenylsilyl]alanine (7c) and 5-{[(Dimethyl)phenylsilyl]methyl}imidazolidine-2,4-dione (8c): From 6c: Analogously to the preparation of 7a and 8a, 6c (0.22 g, 1.0 mmol) was reacted with KOCN (0.10 g, 1.2 mmol) in H₂O/pyridine (1:1, 4 ml) then with aq. HCl soln. (6N, 1 ml) and AcOH (1 ml) to form 8c (0.12 g, 0.49 mmol, 49%). *N*-Carbamoyl amino acid 7c was collected, as described above, after acid treatment before heating to reflux. From 3: Analogously to the preparation of 8a, 3 (5.71 g, 33.6 mmol) was deprotonated with NaOEt (2.65 g, 36.9 mmol) and alkylated with 4c (8.34 g, 30.2 mmol). It was hydrolyzed with NaOH (80 g, 2.0 mol, in 400 ml of H₂O, reflux 48 h), treated with KOCN (1.65 g, 20.5 mmol) in H₂O/pyridine (1:1, 90 ml), then with aq. HCl soln. (6N, 11 ml) and AcOH (11 ml). Recrystallization (H₂O) of the crude product afforded 8c (1.50 g, 6.04 mmol, 20%) as colorless crystals.

Data of **7c**: M.p.: $150.0-150.4^{\circ}$ C (H₂O). IR: 3520s, 3460s, 3360s, 3280s, 3110m, 3060m, 2960m, 2890m, 2710m, 2520m, 1715s, 1660m, 1645m, 1615s, 1565s, 1555s, 1505m, 1485m, 1470w, 1455w, 1430m, 1415m, 1390m, 1340w, 1250s, 1210s, 1170s, 1140m, 1110s, 1070m, 1020w, 1000w, 950w, 920w, 835s, 800m, 775m, 730m, 700s, 675w, 665w. ¹H NMR (DMSO): 7.51-7.48 (m, 2 arom. H); 7.35-7.33 (m, 3 arom. H); 6.14 (d, J = 8.5, HN); 5.54 (s, H_2 N); 4.08 (dt, J = 9.6, 5.5, CH); 1.21, 1.04 (dB of dBX, J_{AB} = 14.6, J_{AX} = 5.3, J_{BX} = 10.1, SiCH₂); 0.25, 0.23 (2s, Me₂Si). ¹³C NMR (DMSO): 176.1, 158.2 (2s, 2 C=O); 138.8 (s, arom. C); 133.5 (d, 2 arom. C); 129.1 (d, arom. C); 127.9 (d, 2 arom. C); 49.4 (d, CH); 19.9 (t, SiCH₂); -2.1, -2.6 (2q, Me₂Si). CI-MS (100°): 284 (5, $[M+NH_4]^+$), 280 (18), 267 (47, $[M+H]^+$), 266 (100, $[M+NH_4-H_2O]^+$), 249 (9, $[M+H-H_2O]^+ \triangleq [8c+H]^+$), 233 (20), 224 (82, $[M+H-CONH]^+ \triangleq [6c+H]^+$), 189 (67), 185 (30), 171 (58).

Data of 8c: M.p.: $160.3-161.2^{\circ}$ C (H₂O). IR: 3220s, 3180s, 3120s, 3060s, 2960m, 2900m, 2740w, 1770s. 1730s, 1695m, 1680w, 1645w, 1635w, 1485w, 1425s, 1415s, 1375m, 1325w, 1305m, 1245m, 1205m, 1195m, 1145w, 1110m, 1090m, 1040m, 1000w, 890w, 835s, 795m, 765m, 740m, 725m, 695m, 630m. ¹H NMR (DMSO): 10.54, 7.74 (2s, br., 2 HN); 7.52-7.49 (m, 2 arom. H); 7.37-7.33 (m, 3 arom. H); 4.03 (dt, J=6.9, 1.5, CH); 1.23, 1.09 (AB of ABX, $J_{AB}=14.9$, $J_{AX}=6.8$, $J_{BX}=7.0$, SiCH₂); 0.31 (s, Me₂Si). ¹³C NMR (DMSO): 177.2, 157.4 (2s, 2 C=O); 138.6 (s, arom. C); 133.5 (d, d arom. C); 129.2 (d, arom. C); 127.9 (d, d).

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arom. C); 55.1 (d, CH); 19.1 (t, SiCH₂); -2.1, -2.2 (2q, Me₂Si). CI-MS: 266 (100, [M + NH₄]+), 249 (12, [M + H]+), 233 (5), 171 (55). Anal. Calcd. for C₁₂H₁₆N₂O₂Si (248.36): C 58.03, H 6.49, N 11.28. Found: C 57.88, H 6.33, N 11.13.

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